



Supplement of

A long-running in situ experiment in clay: 12 years of the Bitumen–Nitrate– Clay interaction experiment at Mont Terri rock laboratory

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Federal Office for the Safety of Nuclear Waste Management

A long running in-situ experiment in clay: **12 years of the Bitumen-Nitrate-Clay** interaction experiment at Mont Terri rock laboratory in Switzerland

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BN: Bitumen-Nitrate-Clay interaction experiment one of the 28 most important and longest-running experiments at Mont Terri rock laboratory



Nitrate plume from bituminized waste => possible perturbations in the near field



Nitrate perturbations studied in BN



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Timeline of injections

| | <mark>2</mark> 011 - 2019: | 2019 – 2023: |
|-------|---|--|
| llay | | |
| nus C | Interval 3: injection of nitrate | Interval 3: selenate, selenate + nitrate |
| Opali | Interval 2: injection of nitrate + acetate | Interval 2: selenate (ongoing) |
| | Interval 1: Injection of nitrate, nitrate + H ₂ | Interval 1: selenate (ongoing) |



Nitrate injection tests

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Nitrate reduction

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- Slow nitrate decrease: mainly diffusion
- Very slow microbial nitrate reduction by clay electron donors: denitrification
- Shift in population towards denitrifiers that are organotrophs



Possible e⁻ donors:

- Most likely: dissolved organic matter (DOM)
- Pyrite FeS₂
- Fe²⁺ containing minerals

100%

50%

0%

90

days days days

0.7

531

Abundancy

Brevundimonas

Pseudomonas

Nitrate reduction with acetate as external electron donor



- 1-2 days lag phase
- Fast microbial nitrate reduction \rightarrow mainly NO₂⁻ production with ~10% denitrification to N₂

 nitrate reducers

Acidovorax: can use Fe(II), but also

acetate as electron donor

Nitrate reduction with H₂ as external electron donor



- 7-10 days lag phase
- Fast nitrate and $pH_2 \downarrow$ due to reduction of NO_3^- by $H_2 \rightarrow NO_2^-$, NH_4^+ , N_2
- Production of ammonium can be biotic or abiotic (Fe surface-catalyzed)

Clostridium

Actinomycetales

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Selenate (+ nitrate) injection tests

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- Long-term: slower selenate removal rate in presence of nitrate, faster in absence of nitrate
- Nitrate suppressed the microbial reduction of selenate

Measurements performed at BRGM by Catherine Lerouge

Fate of selenate in the BN borehole

Solid phase characterized by SEM-EDX and laser ablation coupled with QQQ-ICP-MS



- Selenium was mostly found in pyrite and correlated with sulfur in the sample
- Likely reduced Se species on the solid phase



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BN experiment – what have we learned

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Conclusions: fate of selenate



Modelling in PhreeqC gives good fit

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Slow microbial reaction

 Limited oxidation of clay components expected

> Selenite is formed to some extend

 Selenium species are found on the solid phase, associated with pyrite

Stimulation of microbial activity

- With selenate: sulfate reducing microbes can also reduce selenate, however slowly
- Once nitrate is introduced: shift back to denitrifiers

Further reading

- Bleyen, N., Smets, S., Small, J. *et al.* (2017) Impact of the electron donor on in situ microbial nitrate reduction in Opalinus Clay: results from the Mont Terri rock laboratory (Switzerland). *Swiss Journal of Geosciences*
- Nussbaum, C., Bernier, F., Bleyen, N., *et al.* (2023). 25 years of cross-fertilization between HADES and Mont Terri rock laboratory. *Geological Society, London, Special Publications*
- Bleyen, N., Smets, S., Small, J., *et al.* (2018). Impact of the electron donor on in situ microbial nitrate reduction in Opalinus Clay: results from the Mont Terri rock laboratory (Switzerland). *Mont Terri Rock Laboratory, 20 Years: Two Decades of Research and Experimentation on Claystones for Geological Disposal of Radioactive Waste*
- Bleyen, N., Small, J. S., Mijnendonckx, K., *et. al.* (2021). Ex and in situ reactivity and sorption of selenium in Opalinus clay in the presence of a selenium reducing microbial community. *Minerals*
- Bleyen N., Albrecht A., De Cannière P., et.al. (2019). Non-destructive on-line and long-term monitoring of in situ nitrate and nitrite reactivity in a clay environment at increasing turbidity. Applied Geochemistry

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